

At. No.	Name	Formula	Species Type	Central VE's before bonds	Total VE's after bonds	ED's	Type of Hybrid	$\sigma$ bonds	$\pi$ bonds	Central LP's	Outer Bonds & LP's	Model	Electron Geometry	Molecular Geometry	Polarity Dabyes	Dielectric Constant	mp K	bp K	Molar Mass
1	Hydrogen molecule	H <sub>2</sub>	molecule	1	2	1	s	1	0	0	0	AX	linear	-	13.99	20,271	2,016		
4	Beryllium hydride	BeH <sub>2</sub>	molecule	2	4	2	sp	2	0	0	0	AX <sub>2</sub>	linear	-	523 d	-	11.03		
5	Boron trifluoride	BF <sub>3</sub>	molecule	3	24	3	sp <sup>2</sup>	3	0	0	9	AX <sub>3</sub>	trigonal planar	-	146.3	172.8	67.82		
5	Borane	BH <sub>3</sub>	molecule	3	6	3	sp <sup>2</sup>	3	0	0	0	AX <sub>3</sub>	trigonal planar	-	dimerizes	-	13.83		
5	Diborane	B <sub>2</sub> H <sub>6</sub>	molecule	3	12	3	sp <sup>2</sup>	6	0	0	0	AX <sub>3</sub>	trigonal planar (2 centers)	tetrahedral	0	103.3	180.66	27.67	
5	Ammonia borane	BH <sub>3</sub> NH <sub>3</sub>	molecule	3	14	1	sp <sup>2</sup>	7	0	0	0	AX <sub>3</sub>	trigonal planar (stacked)	trigonal planar	5.2	377	polymerizes	30.87	
6	Carbon monoxide	CO	molecule	4	10	2	sp	1	2	1	1	AXE	linear	0.122	68.13	81.6	28.01		
6	Hydrogen cyanide	HCN	molecule	4	10	2	sp	2	2	0	1	AX <sub>2</sub>	linear	2.98	259.88	299	27,0253		
6	Cyanamide	NCNH <sub>2</sub>	molecule	4	16	2	sp	2	2	0	8	AX <sub>2</sub>	linear	4.27	317	533 d	42.04		
6	Carbon dioxide	CO <sub>2</sub>	molecule	4	16	2	sp	2	2	0	4	AX <sub>2</sub>	linear	-	194,685 s	-	44,009		
6	Isocyanic acid	HNCO	molecule	4	16	2	sp	2	2	0	3	AX <sub>2</sub>	linear	-	187	253 pol.	43.025		
6	Cyanic acid (taut. Isocyanic acid)	HOC(=O)	molecule	4	16	2	sp	2	2	0	8	AX <sub>2</sub>	linear	-	187	253 pol.	43.025		
6	Carbonyl dichloride (Phosgene)	COCl <sub>2</sub>	molecule	4	24	3	sp <sup>2</sup>	3	1	0	8	AX <sub>2</sub>	trigonal planar	1.17	155	281.4	98.91		
6	Methanoic acid, formic acid	HCOOH	molecule	4	16	3	sp <sup>2</sup>	3	1	0	10	AX <sub>2</sub>	trigonal planar	1.41	281.5	373.9	46,025		
6	Ethanoic acid, acetic acid	CH <sub>3</sub> COOH	molecule	4	24	3	sp <sup>2</sup>	3	1	0	16	AX <sub>2</sub>	trigonal planar	1.74	289-290	391-392	60,052		
6	Methanal, formaldehyde	CH <sub>2</sub> O	molecule	4	12	3	sp <sup>2</sup>	3	1	0	2	AX <sub>2</sub>	trigonal planar	2.33	181	254	32,026		
6	Acetone, propanone	(CH <sub>3</sub> ) <sub>2</sub> CO	molecule	4	24	4	sp <sup>2</sup>	3	1	0	8	AX <sub>2</sub>	trigonal planar	2.88	21	178.2	329.23	58.08	
6	Dichloromethane	CH <sub>2</sub> Cl <sub>2</sub>	molecule	4	20	4	sp <sup>2</sup>	4	0	0	6	AX <sub>2</sub>	tetrahedral	1.6	9.1	176.5	312.8	84.93	
6	Ethanol	CH <sub>3</sub> COOH	molecule	4	20	4	sp <sup>3</sup>	4	0	0	6	AX <sub>2</sub>	trigonal planar	1.69	25	155.01	351.38	46,069	
6	Methanol	HCOOH	molecule	4	14	4	sp <sup>3</sup>	4	0	0	5	AX <sub>2</sub>	trigonal planar	1.69	33	175.6	337.8	32,042	
6	Carbon tetrafluoride	CF <sub>4</sub>	molecule	4	32	4	sp <sup>3</sup>	4	0	0	12	AX <sub>2</sub>	tetrahedral	-	85.5	145.3	88,0043		
6	Methane	CH <sub>4</sub>	molecule	4	8	4	sp <sup>3</sup>	4	0	0	0	AX <sub>2</sub>	tetrahedral	-	90,694	111.66	16,043		
6	Oxalic acid	C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	molecule	4	34	3	sp <sup>2</sup>	5	2	0	10	AX <sub>2</sub>	trigonal planar (2 centers)	trigonal planar (2 centers)	462-464	d	90,034		
6	Cyanoacetic acid	C <sub>3</sub> H <sub>3</sub> NO <sub>2</sub>	molecule	4	48	3	sp <sup>2</sup>	9	3	0	12	AX <sub>2</sub>	trigonal planar (3 centers, ring)	trigonal planar (3 centers, ring)	593-633 d	-	125,07		
7	Nitrogen molecule	N <sub>2</sub>	molecule	5	10	2	sp	1	2	1	1	AXE	linear	-	63.23	77,355	28,014		
7	Nitrogen dioxide	NO <sub>2</sub>	molecule	5	17	3	sp <sup>2</sup>	2	1	0.5	5	AX <sub>2</sub> E	trigonal planar	bent	-	263.8	294.3	46,005	
7	Nitrous acid	HNO <sub>2</sub>	molecule	5	18	3	sp <sup>2</sup>	2	1	1	5	AX <sub>2</sub> E	trigonal planar	bent	-	47,013	-	-	
7	Nitrous oxide (dinitrogen monoxide)	N <sub>2</sub> O	molecule	5	16	2	sp	2	2	0	4	AX <sub>2</sub>	linear	0.166	182.29	184.67	44,013		
7	Nitrogen trifluoride	NF <sub>3</sub>	molecule	5	26	4	sp <sup>3</sup>	3	0	1	9	AX <sub>2</sub> E	tetrahedral	0.234	66	144.09	71		
7	Nitrogen trichloride	NCI <sub>3</sub>	molecule	5	26	4	sp <sup>3</sup>	3	0	1	9	AX <sub>2</sub> E	tetrahedral	0.6	233	34	120,36		
7	Ammonia	NH <sub>3</sub>	molecule	5	8	3	sp <sup>3</sup>	3	0	1	0	AX <sub>2</sub> E	tetrahedral	1.42	195.42	239.81	17,031		
7	Nitric acid	HNO <sub>3</sub>	molecule	5	24	3	sp <sup>2</sup>	3	1	0	8	AX <sub>2</sub>	trigonal planar	2.17	231 (pure)	356 (pure)	63,012		
7	Dinitrogen tetroxide	N <sub>2</sub> O <sub>4</sub>	molecule	5	34	4	sp <sup>2</sup>	6	2	0	9	AX <sub>2</sub>	trigonal planar (2 centers)	tetrahedral pyramidal (2 centers)	-0	261.9	294.84	92.01	
8	Oxygen molecule	O <sub>2</sub>	molecule	6	12	2	sp	1	1	2	2	AX	trigonal planar	linear	-	54.36	90,188	31,998	
8	Diethyl ether	C <sub>4</sub> H <sub>10</sub> O	molecule	6	32	4	sp <sup>3</sup>	2	0	2	12	AX <sub>2</sub>	tetrahedral	bent	1.15	4.3	156.8	307.8	74,123
8	Water	H <sub>2</sub> O	molecule	6	8	4	sp <sup>3</sup>	2	0	2	0	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	1,8546	80	273.15	373.13	18,015
8	Ozone	O <sub>3</sub>	molecule	6	18	3	sp <sup>2</sup>	2	1	1	5	AX <sub>2</sub> E <sub>2</sub>	trigonal planar	bent	0.53	81	161	47,997	
8	Hydrogen peroxide	H <sub>2</sub> O <sub>2</sub>	molecule	6	14	1	sp <sup>3</sup>	3	0	2	2	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	2.26	272.2	423.3 (d)	34,014	
9	Hydrogen fluoride	HF	molecule	7	8	4	sp <sup>3</sup>	1	0	3	0	AX <sub>3</sub>	tetrahedral	linear	1.86	189.6	292.6	20,006	
9	Fluorine molecule	F <sub>2</sub>	molecule	7	14	4	sp <sup>3</sup>	1	0	3	3	AX <sub>3</sub>	tetrahedral	linear	-	53.48	85.03	37,996	
13	Aluminum bromide	AlBr <sub>3</sub>	molecule	3	24	3	sp <sup>2</sup>	3	0	0	9	AX <sub>2</sub>	trigonal planar	-	370.5	528	266,694		
13	Aluminum hydride	AlH <sub>3</sub>	molecule	3	6	3	sp <sup>2</sup>	3	0	0	0	AX <sub>2</sub>	trigonal planar	-	423 (378)	-	30,006		
14	Silane	SiH <sub>4</sub>	molecule	4	8	4	sp <sup>3</sup>	4	0	0	0	AX <sub>2</sub>	tetrahedral	0	88.1	161.2	32,117		
14	Silicon tetrachloride	SiCl <sub>4</sub>	molecule	4	32	4	sp <sup>3</sup>	4	0	0	12	AX <sub>2</sub>	tetrahedral	-	204,41	330.8	169.9		
15	Phosphine	PH <sub>3</sub>	molecule	5	8	3	sp <sup>3</sup>	3	0	1	0	AX <sub>2</sub> E	tetrahedral	0.58	140.3	185.5	33,99758		
15	Phosphorous oxychloride	POCl <sub>3</sub>	molecule	5	32	4	sp <sup>3</sup>	4	1	0	11	AX <sub>2</sub>	tetrahedral	2.54	274.4	378.9	153,32		
15	Phosphorus pentachloride	PF <sub>5</sub>	molecule	5	40	5	sp <sup>3</sup> d	5	0	0	15	AX <sub>2</sub>	trigonal bipyramidal	0	173,37	188.6	125,9657776		
16	Hydrogen sulfide	H <sub>2</sub> S	molecule	6	8	4	sp <sup>3</sup>	2	0	2	0	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	0.97	187.7	213.6	34,08	
16	Sulfur dichloride	SCl <sub>2</sub>	molecule	6	20	4	sp <sup>3</sup>	2	0	2	8	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	152.2	332 d	102,96		
16	Sulfur dioxide	SO <sub>2</sub>	molecule	6	18	3	sp <sup>2</sup>	2	2	1	4	AX <sub>2</sub> E	trigonal planar	bent	1.62	201	263	64,066	
16	Sulfur trioxide	SO <sub>3</sub>	molecule	6	24	3	sp <sup>2</sup>	3	1	0	8	AX <sub>2</sub>	trigonal planar	-	290	318	80,056		
16	Dimethyl sulfoxide, DMSO	(CH <sub>3</sub> ) <sub>2</sub> SO	molecule	6	26	4	sp <sup>3</sup>	3	1	1	8	AX <sub>2</sub> E	tetrahedral	3.98	49	292	462	78.13	
16	Sulfur tetrafluoride	SF <sub>4</sub>	molecule	6	34	5	sp <sup>3</sup> d	4	0	2	12	AX <sub>2</sub> E	trigonal bipyramidal	see-saw	0.532	152.15	235.15	108.07	
16	Sulfuryl chloride	SO <sub>2</sub> Cl <sub>2</sub>	molecule	6	32	4	sp <sup>3</sup>	4	2	0	10	AX <sub>2</sub>	tetrahedral	tetrahedral	219.1	342.5	134,9688		
16	Sulfur hexafluoride	SF <sub>6</sub>	molecule	6	48	6	sp <sup>3</sup> d <sup>2</sup>	6	0	0	18	AX <sub>2</sub>	octahedral	0	209	222.3	146,05		
17	Chlorine monofluoride	ClF	molecule	7	14	4	sp <sup>3</sup>	1	0	3	3	AX <sub>2</sub> E <sub>3</sub>	tetrahedral	linear	0.881	117.5	173.1	54,45	
17	Hydrogen chloride	HCl	molecule	7	8	4	sp <sup>3</sup>	1	0	3	0	AX <sub>2</sub> E <sub>3</sub>	tetrahedral	linear	1.05	158.93	188.1	36,46	
17	Chlorine molecule	Cl <sub>2</sub>	molecule	7	14	4	sp <sup>3</sup>	1	0	3	3	AX <sub>2</sub> E <sub>3</sub>	tetrahedral	linear	-	171.6	239.11	70.9	
24	Chromium trioxide	CrO <sub>3</sub>	molecule	6	24	4	sp <sup>3</sup>	3	3	0	6	AX <sub>2</sub>	tetrahedral	trigonal planar	-	470	523 d	99,993	
33	Arsine	AsH <sub>3</sub>	molecule	5	8	3	sp <sup>3</sup>	3	0	1	0	AX <sub>2</sub> E	tetrahedral	trigonal pyramidal	0.2	162	210.7	77,8454	
33	Arsenic pentafluoride	AsF <sub>5</sub>	molecule	5	38	5	sp <sup>3</sup> d	5	0	0	15	AX <sub>2</sub>	trigonal bipyramidal	trigonal bipyramidal	-	193.2	220.2	169,9136	
34	Hydrogen selenide	H <sub>2</sub> Se	molecule	6	8	4	sp <sup>3</sup>	2	0	2	0	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	207,42	231.9	80,98		
34	Selenium dichloride	SeCl <sub>2</sub>	molecule	6	20	4	sp <sup>3</sup>	2	0	2	8	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	-	319-320	-	148,87	
34	Selenium dibromide	SeBr <sub>2</sub>	molecule	6	20	4	sp <sup>3</sup>	2	0	2	8	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	-	238,99	-	-	
34	Selenium hexafluoride	SeCl <sub>6</sub>	molecule	6	48	6	sp <sup>3</sup> d <sup>2</sup>	6	0	0	18	AX <sub>2</sub>	octahedral	0	234	238.7 s	192,9534		
35	Hydrogen bromide	HBr	molecule	7	8	4	sp <sup>3</sup>	1	0	3	0	AX <sub>2</sub> E <sub>3</sub>	tetrahedral	linear	0.82	186.2	205.3	80,91	
35	Bromine molecule	Br <sub>2</sub>	molecule	7	14	4	sp <sup>3</sup>	1	0	3	3	AX <sub>2</sub> E <sub>3</sub>	tetrahedral	linear	-	265.8	332	159,808	
52	Hydrogen telluride	H <sub>2</sub> Te	molecule	6	8	4	sp <sup>3</sup>	2	0	2	0	AX <sub>2</sub> E <sub>2</sub>	tetrahedral	bent	224	-	270.9	129,6158	
53	Hydrogen iodide	HI	molecule	7	8	4	sp <sup>3</sup>	1	0	3	0	AX <sub>2</sub> E <sub>3</sub>	tetrahedral	linear	0.38	222.35	237.79	127,912	
53	Iodine molecule	I <sub>2</sub>	molecule	7	14	4	sp <sup>3&lt;/sup</sup>												